

Sc₂MgGa₂ and Y₂MgGa₂

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Received 16 September 2008

Accepted 7 January 2009

Online 7 February 2009

Scandium magnesium gallide, Sc₂MgGa₂, and yttrium magnesium gallide, Y₂MgGa₂, were synthesized from the corresponding elements by heating under an argon atmosphere in an induction furnace. These intermetallic compounds crystallize in the tetragonal Mo₂FeB₂-type structure. All three crystallographically unique atoms occupy special positions and the site symmetries of (Sc/Y, Ga) and Mg are *m2m* and *4/m*, respectively. The coordinations around Sc/Y, Mg and Ga are pentagonal (Sc/Y), tetragonal (Mg) and triangular (Ga) prisms, with four (Mg) or three (Ga) additional capping atoms leading to the coordination numbers [10], [8+4] and [6+3], respectively. The crystal structure of Sc₂MgGa₂ was determined from single-crystal diffraction intensities and the isostructural Y₂MgGa₂ was identified from powder diffraction data.

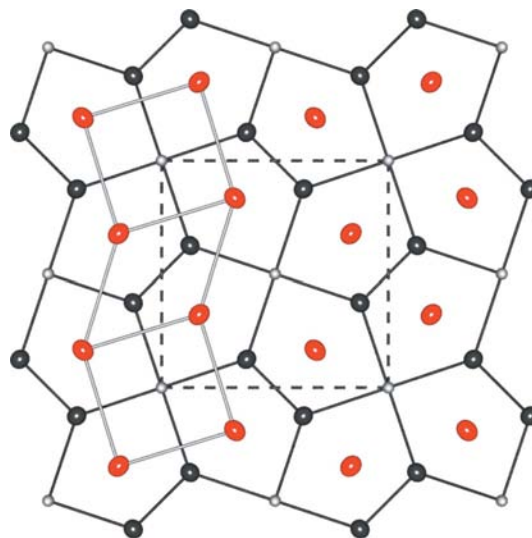
Comment

The potential use of magnesium alloys as storage materials for hydrogen has led to a large number of investigations on different magnesium alloys (Selvam *et al.*, 1986; Sahlberg & Andersson, 2007; Sahlberg *et al.*, 2007; Zlotea *et al.*, 2008). Different compositions in the ternary Sc–Mg–Ga and Y–Mg–Ga systems have been synthesized in order to investigate the hydrogen absorption properties of these compounds, and several intermediate phases have been found. There are numerous RE₂T₂X compounds (where RE is a rare earth metal, T is a transition metal and X is a p-block metal) which often crystallize in the W₂CoB₂-type (Rieger *et al.*, 1966) or Mo₂FeB₂-type structures (Rieger *et al.*, 1964). These types of compounds were reviewed by Lukachuk & Pöttgen (2003). The Mo₂FeB₂-type structure is a ternary ordered version of the U₃Si₂-type structure (Zachariasen, 1949), where the two uranium sites are occupied by different atoms.

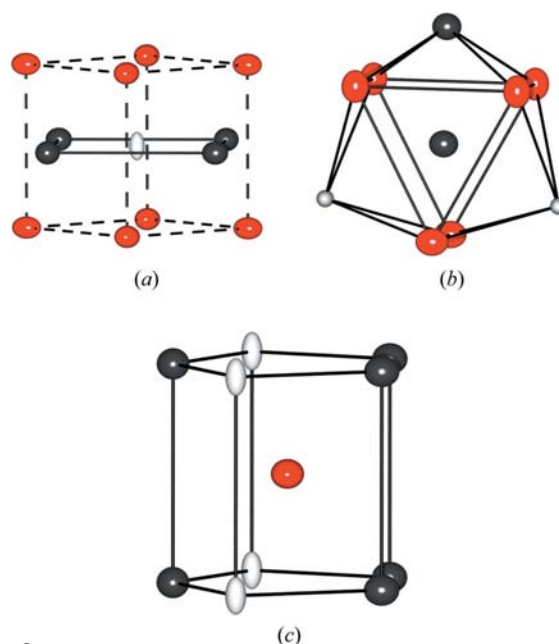
We have synthesized single crystals of Sc₂MgGa₂, determined the crystal structure and found it to have the Mo₂FeB₂-type structure. Prior to this investigation, there were no ternary gallide phases reported with the Mo₂FeB₂-type structure, only for the binary Ce₃Ga₂ and Gd₃Ga₂ structures. It should also be noted that in the Sc–Mg–Ga ternary system this is, to our knowledge, the first reported ternary phase.

The Mo₂FeB₂-type structure, space group *P4/mbm*, can be described as two intergrown slabs of CsCl and AIB₂ types, as shown in Fig. 1. The compositions in the CsCl and AIB₂ slabs are ScMg and ScGa₂, respectively. The structure is layered in the *z* direction, and in Sc₂MgGa₂ the Mg and Ga atoms form a Ga₂Mg layer at *z* = 0 and the Sc atoms form a layer at *z* = 0.5.

The coordination around Mg is approximately a tetragonal prism of eight Sc atoms with four additional Ga atoms capping

**Figure 1**

The crystal structure of Sc₂MgGa₂, viewed down the *c* axis, showing the Ga₂Mg and Sc layers. The two slabs of 'CsCl' and 'AIB₂' are emphasized. Key: Mg white, Ga black and Sc light grey (red in the electronic version of the paper). Displacement ellipsoids are drawn at the 95% probability level.

**Figure 2**

Coordination around (a) the Mg atom, (b) the Ga atom and (c) the Sc atom. Displacement ellipsoids are drawn at the 95% probability level.

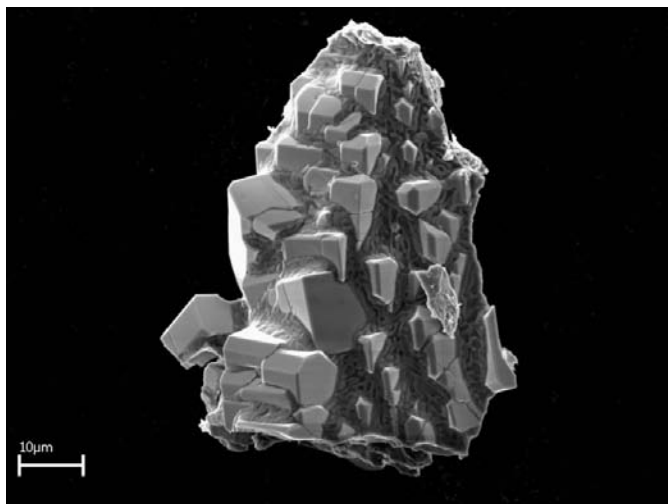


Figure 3
SEM image of a sample with the overall composition ScMg_4Ga . The Sc_2MgGa_2 crystals are shown in the matrix.

the rectangular faces (Fig. 2a). Ga is coordinated by nine atoms in a capped triangular prismatic arrangement, *viz.* six Sc atoms in the prism plus two Mg and one Ga atom outside the rectangular faces (Fig. 2b). Sc is surrounded by two $[\text{Ga}_3\text{Mg}_2]$ rings, forming a distorted pentagonal prism (Fig. 2c).

The interatomic distances are in agreement with the corresponding binary compounds. The shortest interatomic distance is the Ga··Ga distance [2.535 (2) Å], which is slightly longer than the interatomic distance (2.48 Å) in orthorhombic α -Ga. The bonding in the compound is believed to be metallic.

The unit cell for the isostructural Y_2MgGa_2 compound was determined to be $a = 7.428$ (2) Å and $c = 4.2537$ (2) Å.

Experimental

Appropriate amounts of the elements (Mg 99.99%, Sc 99.95%, Ga 99%) were melted inside a tantalum tube sealed under an argon atmosphere, using a high-frequency induction furnace. The tubes were heated to ~ 1373 K for 10 min, and then cooled to room temperature. Large single crystals of Sc_2MgGa_2 were obtained on the surface of magnesium-rich samples. The mm-sized single crystals were cut into smaller pieces. The bulk samples were characterized by X-ray powder diffraction. The chemical composition of the single crystals was also analysed with SEM-EDS (scanning electron microscopy–energy-dispersive X-ray spectroscopy). The image in Fig. 3 is taken from a sample with the nominal composition ScMgGa . The composition from EDS was found to be $\text{Sc}_{0.37}\text{Mg}_{0.20}\text{Ga}_{0.43}$ after correction by the ZAF method, which corrects for atomic number (Z), absorption (A) and fluorescence (F). Y_2MgGa_2 was synthesized using the same method as for Sc_2MgGa_2 . The phase was found in a multiphase sample with the overall composition YMgGa . Attempts to synthesize large single crystals were not successful, probably due to the high stability of YMgGa and YGa_2 . The crystal structure was indexed from powder X-ray diffraction data. The unit cell was determined using the *CHECKCELL* program (Laguier & Bochu, 2004).

Table 1
Selected bond lengths (Å).

Sc1–Ga1 ⁱ	2.8286 (15)	Ga1–Ga1 ⁱⁱⁱ	2.535 (2)
Sc1–Mg1 ⁱ	3.2931 (6)	Ga1–Mg1 ^{iv}	2.8283 (6)
Sc1–Sc1 ⁱⁱ	3.505 (4)		

Symmetry codes: (i) $x, y, z + 1$; (ii) $-x, -y + 1, -z + 1$; (iii) $-x + 1, -y + 2, -z$; (iv) $-x + \frac{1}{2}, y - \frac{1}{2}, -z$.

Crystal data

Sc_2MgGa_2	$Z = 2$
$M_r = 253.67$	Mo $K\alpha$ radiation
Tetragonal, $P4/mbm$	$\mu = 16.43 \text{ mm}^{-1}$
$a = 7.1577$ (10) Å	$T = 293$ (2) K
$c = 3.9166$ (8) Å	$0.14 \times 0.09 \times 0.04 \text{ mm}$
$V = 200.66$ (6) Å ³	

Data collection

Bruker APEX-I diffractometer	2538 measured reflections
Absorption correction: multi-scan (<i>SADABS</i> ; Sheldrick, 2004)	154 independent reflections
$T_{\min} = 0.20, T_{\max} = 0.52$	135 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.069$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.028$	11 parameters
$wR(F^2) = 0.069$	$\Delta\rho_{\text{max}} = 0.84 \text{ e \AA}^{-3}$
$S = 1.16$	$\Delta\rho_{\text{min}} = -0.73 \text{ e \AA}^{-3}$
154 reflections	

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINT* (Bruker, 2001); data reduction: *SAINT* (Bruker, 2001); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *DIAMOND* (Brandenburg, 2007); software used to prepare material for publication: *SHELXL97* (Sheldrick, 2008) and *pubCIF* (Westrip, 2009).

The authors thank the Swedish Research Council and the Royal Swedish Academy of Sciences for financial support.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: BD3001). Services for accessing these data are described at the back of the journal.

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